



## TITANIUM OXIDES – ELECTRICAL ALL-ROUNDERS

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At Fraunhofer IKTS, titanium oxides are adapted to customer-specific user requirements, components are manufactured from them and extensively tested. Titanium oxides are characterized by extraordinary variability concerning their oxide compounds. Besides titanium dioxide ( $\text{TiO}_2$ ), which is used for numerous products, such as wall paints, cosmetic products, paper or carriers of catalysts, there are multiple other oxide compositions of titanium. Their variable compositions range from  $\text{Ti}_2\text{O}$  to  $\text{TiO}_2$ . Even a tiny oxygen deficiency in  $\text{TiO}_2$  can decrease the electrical resistance from  $10^{12} \Omega\text{cm}$  to a level between  $10^5$  and  $10^3 \Omega\text{cm}$ . The crystallographic phase change to e.g.  $\text{Ti}_4\text{O}_7$  shifts the electrical resistance to  $10^{-2}$ – $10^{-3} \Omega\text{cm}$ . Figure 1 illustrates the proportions across 15 orders of magnitude of electrical resistivity covered by various titanium oxide types. Furthermore, non-linear correlation of voltage and current can be created through the doping of  $\text{TiO}_2$ . Electrical permittivity is particularly high, at a level between 60–800 depending on the applied frequency. Significant oxygen conductivity already sets in at  $500^\circ\text{C}$ . Also of technical interest is the photocatalytic effect, which is particularly linked with the  $\text{TiO}_2$  type anatase.

### Electro-technical applications

This wide range of properties offers advantages in numerous technical applications: specific materials for electrodes, thermoelectric materials, varistor materials, active catalyst substances, electrical conductors, semi-conductors and insulators or even sensors for oxygen identification and analysis. These applications have a variety of different requirements regarding densification, mechanical strength or oxidation stability. Through the specific modification of the manufacturing parameters, titanium oxide

ceramics can be produced as dense and high-strength material or made to be porous and gas-permeable.  $\text{TiO}_2$  is completely oxidation-resistant and can be used up to  $1500^\circ\text{C}$  under air. Suboxides are oxidation-resistant up to  $400^\circ\text{C}$ . Furthermore, titanium oxides are chemically stable against almost all reactive substances.

The electrical parameters can be modified to suit almost all applications. Joined compounds of titanium dioxide and titanium suboxides can be produced which combine electrically insulating and electrically conductive parts within a monolithic component (Figure 2).

### Services offered

- Adaptation of titanium oxides to specific application requirements
- Manufacture of customer-specific titanium oxide components
- Implementation of material- and application-orientated investigations and tests

1 Illustration of the multiple specific electrical resistivities of titanium oxides at room temperature.

2 Monolithic joint consisting of  $\text{TiO}_2$  (shell) and  $\text{Ti}_4\text{O}_7$  (core).